

Electricity Review

Electrons

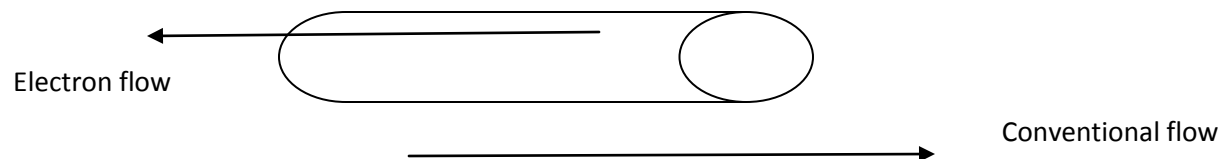
- electrons are what is actually moving through a conductor.
- one electron is a very small, so we 'bundle' them into 'coulombs'.
1 coulomb \ominus = 6.24×10^{18} electrons
- 1 electron has the charge of $1 / 6.24 \times 10^{18}$ coulombs
... or... 1 electron has a charge of $-1.6 \times 10^{-19} \text{ C}$
It's negative because an electron is negatively charged.

$e = -1.6 \times 10^{-19} \text{ C}$ $1 \text{ coulomb} = 6.24 \times 10^{18} \text{ electrons}$
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put these on formula sheet!

Circuits

- need a full, closed circuit (circle) in order for electricity to flow.
- know circuit symbols for: battery (+/- end as well), resistance, switch (open/closed) and lightbulb.
- early in the study of electricity, Benjamin Franklin proposed that positiveness flowed. This idea 'stuck' for quite a while. So although we KNOW electrons (-ve flow) move, we sometimes refer to +ve flow and we call this 'conventional current'.
- conventional current = +ve flow (= opposite direction to electron flow)



Simple formulas

$Q = Ne$ where Q = charge in coulombs (C)

N = number of electrons. + N = excess electrons

- N = deficit of electrons

$$e = -1.6 \times 10^{-19} \text{ C}$$

ie: What is the total charge on an object that has a deficit of 3 electrons?

$$Q = ? \quad N = -3 \quad e = -1.6 \times 10^{-19} \text{ C}$$

$$Q = Ne = (-3) (-1.6 \times 10^{-19} \text{ C}) = 4.8 \times 10^{-19} \text{ C}$$

The charge is $4.8 \times 10^{-19} \text{ C}$

$I = Q/t$ where $I = \text{current (C/s)}$ * $1 \text{ C/s} = 1 \text{ ampere or 1 amp (A)}$

$t = \text{time (s)}$ $Q = \text{charge as above (C)}$

Current is the rate of electron flow. We don't count 'electrons' per say, but coulombs of electrons. Remember that electrons are very very small so we bundle them.

$V = E/Q$ where $V = \text{potential difference (V)}$

$E = \text{energy (J)}$ $Q = \text{charge as above (C)}$

Potential difference (more commonly known as 'voltage') measures the energy each coulomb of electrons possesses. The more energy a coulomb of electrons has, the more work they can do!

These formulas are pretty straight forward. To master your understanding, practice by doing the following problems:

Homework:

Copy: The 3 formulas onto your formula sheet

Problems:

11.5 - p. 518 # 1,2,3,4,5

11.3 - p. 513 # 1,2,3,4,6,7,8 (no #5)